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APPLICATION NO.	FILING DATE	FIRST NAMED INVENTOR	ATTORNEY DOCKET NO.	CONFIRMATION NO.
10/822,016	04/08/2004	Chris M. Carlson	MI22-2449	1689
21567 7590 09/04/2008 WELLS ST. JOHN P.S. 601 W. FIRST AVENUE, SUITE 1300 SPOKANE, WA 99201				
EXAMINER VETTER, ROBERT A				
ART UNIT		PAPER NUMBER		
1792				
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**Please find below and/or attached an Office communication concerning this application or proceeding.**

The time period for reply, if any, is set in the attached communication.

# Office Action Summary

**Application No.**

10/822,016

**Applicant(s)**

CARLSON ET AL.

**Examiner**

ROBERT VETERE

**Art Unit**

1792

-- The MAILING DATE of this communication appears on the cover sheet with the correspondence address --  
**Period for Reply**

A SHORTENED STATUTORY PERIOD FOR REPLY IS SET TO EXPIRE 3 MONTH(S) OR THIRTY (30) DAYS, WHICHEVER IS LONGER, FROM THE MAILING DATE OF THIS COMMUNICATION.

- Extensions of time may be available under the provisions of 37 CFR 1.136(a). In no event, however, may a reply be timely filed after SIX (6) MONTHS from the mailing date of this communication.
- If NO period for reply is specified above, the maximum statutory period will apply and will expire SIX (6) MONTHS from the mailing date of this communication.
- Failure to reply within the set or extended period for reply will, by statute, cause the application to become ABANDONED (35 U.S.C. § 133). Any reply received by the Office later than three months after the mailing date of this communication, even if timely filed, may reduce any earned patent term adjustment. See 37 CFR 1.704(b).

**Status**

- 1) ☒ Responsive to communication(s) filed on 21 July 2008.  
2a) ☐ This action is **FINAL**. 2b) ☒ This action is non-final.  
3) ☐ Since this application is in condition for allowance except for formal matters, prosecution as to the merits is closed in accordance with the practice under *Ex parte Quayle*, 1935 C.D. 11, 453 O.G. 213.

**Disposition of Claims**

- 4) ☒ Claim(s) 12-15, 20, 28 and 87 is/are pending in the application.  
4a) Of the above claim(s) \_\_\_\_\_ is/are withdrawn from consideration.  
5) ☐ Claim(s) \_\_\_\_\_ is/are allowed.  
6) ☒ Claim(s) 12-15, 20, 28, 87 is/are rejected.  
7) ☐ Claim(s) \_\_\_\_\_ is/are objected to.  
8) ☐ Claim(s) \_\_\_\_\_ are subject to restriction and/or election requirement.

**Application Papers**

- 9) ☐ The specification is objected to by the Examiner.  
10) ☐ The drawing(s) filed on \_\_\_\_\_ is/are: a) ☐ accepted or b) ☐ objected to by the Examiner.  
Applicant may not request that any objection to the drawing(s) be held in abeyance. See 37 CFR 1.85(a).  
Replacement drawing sheet(s) including the correction is required if the drawing(s) is objected to. See 37 CFR 1.121(d).  
11) ☐ The oath or declaration is objected to by the Examiner. Note the attached Office Action or form PTO-152.

**Priority under 35 U.S.C. § 119**

- 12) ☐ Acknowledgment is made of a claim for foreign priority under 35 U.S.C. § 119(a)-(d) or (f).  
a) ☐ All b) ☐ Some \* c) ☐ None of:  
1. ☐ Certified copies of the priority documents have been received.  
2. ☐ Certified copies of the priority documents have been received in Application No. \_\_\_\_\_.  
3. ☐ Copies of the certified copies of the priority documents have been received in this National Stage application from the International Bureau (PCT Rule 17.2(a)).

\* See the attached detailed Office action for a list of the certified copies not received.

**Attachment(s)**

- 1) ☒ Notice of References Cited (PTO-892)  
2) ☐ Notice of Draftsperson's Patent Drawing Review (PTO-948)  
3) ☐ Information Disclosure Statement(s) (PTO-8508)  
Paper No(s)/Mail Date \_\_\_\_\_  
4) ☐ Interview Summary (PTO-413)  
Paper No(s)/Mail Date \_\_\_\_\_  
5) ☐ Notice of Informal Patent Application  
6) ☐ Other: \_\_\_\_\_

## DETAILED ACTION

### *Continued Examination Under 37 CFR 1.114*

1. A request for continued examination under 37 CFR 1.114, including the fee set forth in 37 CFR 1.17(e), was filed in this application after final rejection. Since this application is eligible for continued examination under 37 CFR 1.114, and the fee set forth in 37 CFR 1.17(e) has been timely paid, the finality of the previous Office action has been withdrawn pursuant to 37 CFR 1.114. Applicant's submission filed on 7/21/2008 has been entered.

### *Claim Rejections - 35 USC § 103*

2. The following is a quotation of 35 U.S.C. 103(a) which forms the basis for all obviousness rejections set forth in this Office action:

(a) A patent may not be obtained though the invention is not identically disclosed or described as set forth in section 102 of this title, if the differences between the subject matter sought to be patented and the prior art are such that the subject matter as a whole would have been obvious at the time the invention was made to a person having ordinary skill in the art to which said subject matter pertains. Patentability shall not be negated by the manner in which the invention was made.

3. Claims 14-15 are rejected under 35 U.S.C. 103(a) as being unpatentable over Lee et al. (2004/0238872, hereinafter "Lee") in light of Lee et al. (US 7,151,039, hereinafter "Lee II") and further in light of Yamamoto (US 6,936,901) and further in light of Ahn et al. (US 7,135,421).

**Claims 14-15:** Lee teaches a method of forming a high-k dielectric layer comprising hafnium aluminum oxide using ALD (¶ 0043), wherein a silicon substrate is loaded on a wafer stage of an ALD apparatus (¶ 0043), a pulse of a first precursor containing a first element (e.g., Hf) is supplied and chemisorbed onto the substrate (¶ 0043) to form a monolayer (¶ 0043, Fig. 6), the part of the first precursor not absorbed is exhausted from the apparatus (¶ 0043), a pulse of a second precursor containing a second element (e.g., Al) is supplied and chemisorbed into the first element of the first precursor (¶ 0043, Figs. 7-8), the part of the first precursor not absorbed is exhausted from the apparatus (cl. 3), providing a reactant (e.g., Ozone) which reacts with both the first and second elements to form a high-k dielectric layer that contains both the first and second elements (¶ 0046) and exhausting the ALD apparatus to remove the excess reactant (cl. 2).

Art Unit: 1792

Lee also teaches that the Al-containing precursor is TMA (¶ 0043) and that the oxidant is  $O_3$  (¶ 0046). What it does not teach is that the Hf-containing precursor is TMEAH or TDMAH. Lee II teaches a method of forming an oxide layer using ALD (Col. 2: 58-67) containing, for example, Al and/or Hf (3:9-11), wherein the hafnium-containing precursor is TDMAH (i.e.  $Hf[N(C_2H_5)_2]_4$ ) or tetrakis-ethylmethylamino hafnium (i.e.  $Hf[NC_2H_5CH_3]_4$ , claimed TMEAH) (3:12-15). Furthermore, the selection of a known material based on its suitability for its intended use supported a prima facie obviousness determination in *Sinclair & Carroll Co. v. Interchemical Corp.*, 65 USPQ 297 (1945). Thus, it would have been obvious to one of ordinary skill in the art at the time the invention was made to have used either TDMAH or TMEAH as the hafnium-containing precursor in the method of Lee with the predictable expectation of successfully forming a high-k dielectric layer containing hafnium and aluminum.

Lee also fails to teach that the aluminum-containing precursor is provided before the hafnium-containing precursor. Yamamoto teaches a method of forming oxide layers of aluminum (8: 43-60) and hafnium (8:61-9:8) using ALD (see, e.g., 8:20) wherein the aluminum is introduced before the hafnium. Furthermore, the teaching of Lee is not confined to introducing hafnium before aluminum because Lee teaches only teaches that first precursor contains hafnium e.g. (see ¶ 0043, line 5) and likewise with the second precursor (see ¶ 0043, lines 13-14). Thus, because both Lee and Yamamoto teach methods of using ALD to form oxide layers comprising aluminum and hafnium, it would have been obvious to one of ordinary skill in the art at the time the invention was made to have introduced aluminum before hafnium, as taught by Yamamoto, with the predictable expectation of successfully forming a high-k dielectric layer.

Lee also fails to teach that the pulse sequence of  $TMA-(TMEAH-O_3)_x$ , where x is at least two. Ahn teaches a method of forming a layer of hafnium aluminum oxide using ALD and also that in order to control the dielectric constant of the layer, it is best to control the number of cycles of hafnium-reactant relative to the number of cycles of aluminum (13:20-44). "[W]here the general conditions of a claim are disclosed in the prior art, it is not inventive to discover the optimum or workable ranges by routine experimentation." *In re Aller*, 105 USPQ 233, 235 (CCPA 1955). Thus, it would have been obvious to one of ordinary skill in the art at the time the invention was made to have controlled the number of cycles of

Art Unit: 1792

TMEA-H-O<sub>3</sub> using, for example, 2 or more sequences, in order to obtain the desired dielectric constant in that layer.

4. Claims 12-13, 28, 28 and 87 are rejected under 35 U.S.C. 103(a) as being unpatentable over Lee, in light of Lee II and further in light of Yamamoto.

**Claims 12-13, 20, 28 and 87:** Lee teaches a method of forming a high-k dielectric layer comprising hafnium aluminum oxide using ALD (¶¶ 0043), wherein a silicon substrate is loaded on a wafer stage of an ALD apparatus (¶¶ 0043), a pulse of a first precursor containing a first element (e.g., Hf) is supplied and chemisorbed onto the substrate (¶¶ 0043) to form a monolayer (¶¶ 0043, Fig. 6), the part of the first precursor not absorbed is exhausted from the apparatus (¶¶ 0043), a pulse of a second precursor containing a second element (e.g., Al) is supplied and chemisorbed into the first element of the first precursor (¶¶ 0043, Figs. 7-8), the part of the first precursor not absorbed is exhausted from the apparatus (cl. 3), providing a reactant (e.g., Ozone) which reacts with both the first and second elements to form a high-k dielectric layer that contains both the first and second elements (¶¶ 0046) and exhausting the ALD apparatus to remove the excess reactant (cl. 2).

Lee also teaches that the Al-containing precursor is TMA (¶¶ 0043) and that the oxidant is O<sub>3</sub> (¶¶ 0046). What it does not teach is that the Hf-containing precursor is TMEA-H or TDMA-H. Lee II teaches a method of forming an oxide layer using ALD (Col. 2: 58-67) containing, for example, Al and/or Hf (3:9-11), wherein the hafnium-containing precursor is TDMA-H (i.e. Hf[N(C<sub>2</sub>H<sub>5</sub>)<sub>2</sub>]<sub>4</sub>) or tetrakis-ethylmethylamino hafnium (i.e. Hf[NC<sub>2</sub>H<sub>5</sub>CH<sub>3</sub>]<sub>4</sub>, claimed TMEA-H) (3:12-15). Furthermore, the selection of a known material based on its suitability for its intended use supported a prima facie obviousness determination in *Sinclair & Carroll Co. v. Interchemical Corp.*, 65 USPQ 297 (1945). Thus, it would have been obvious to one of ordinary skill in the art at the time the invention was made to have used either TDMA-H or TMEA-H as the hafnium-containing precursor in the method of Lee with the predictable expectation of successfully forming a high-k dielectric layer containing hafnium and aluminum.

Lee also fails to teach that the aluminum-containing precursor is provided before the hafnium-containing precursor. Yamamoto teaches a method of forming oxide layers of aluminum (8: 43-60) and hafnium (8:61-9:8) using ALD (see, e.g., 8:20) wherein the aluminum is introduced before the hafnium.

Art Unit: 1792

Furthermore, the teaching of Lee is not confined to introducing hafnium before aluminum because Lee teaches only teaches that first precursor contains hafnium e.g. (see ¶ 0043, line 5) and likewise with the second precursor (see ¶ 0043, lines 13-14). Thus, because both Lee and Yamamoto teach methods of using ALD to form oxide layers comprising aluminum and hafnium, it would have been obvious to one of ordinary skill in the art at the time the invention was made to have introduced aluminum before hafnium, as taught by Yamamoto, with the predictable expectation of successfully forming a high-k dielectric layer.

The combination of Lee, Lee II and Yamamoto, however, fails to expressly teach that the utilization of TMA improves conformatlity relative to the conformatlity that would exist in the absence of TMA. This limitation, however, is inherently contained in these references because the method taught by the combination of Lee, Lee II and Yamamoto is identical to the method claimed by applicants in claims 20 and 87.

#### ***Response to Arguments***

5. Applicant's arguments filed 7/21/2008 have been fully considered but they are not persuasive.

Applicant first argues that the amendment to claims 14 and 15 places those claims in condition for allowance. In view of the new rejection above, this argument is moot.

Applicant next argues that the limitation regarding conformality in claims 20 and 87 is not inherent in either Lee I or the combination of Lee I, Lee II and Yamamoto. This is not persuasive. While it is not inherent in Lee I on its own, the rejection is based on the combined method of Lee I, Lee II and Yamamoto. This combination of references teaches all the limitation of claims 20 and 87 and therefore inherently will improve conformality by using TMA. The fact that none of the references provide motivation to use TMA as a means of improving conformality is not relevant because the fact that applicant has recognized another advantage which would flow naturally from following the suggestion of the prior art cannot be the basis for patentability when the differences would otherwise be obvious. See *Ex parte Obiaya*, 227 USPQ 58, 60 (Bd. Pat. App. & Inter. 1985). Finally, inherency does not require a disclosure of the benefit conferred, as applicant suggests on page 12 of applicant's arguments.

Applicant also argues that "the relative concentrations of TMA to TMEAH (or TDMAH) may affect whether the process actually improves conformality relative to a process lacking TMA" (Applicant's

Art Unit: 1792

arguments, p. 12). There is not persuasive. This statement is not reflected in the new claim language added in this amendment.

***Conclusion***

Any inquiry concerning this communication or earlier communications from the examiner should be directed to ROBERT VETERE whose telephone number is (571)270-1864. The examiner can normally be reached on Mon-Fri 9-6.

If attempts to reach the examiner by telephone are unsuccessful, the examiner's supervisor, Michael Cleveland can be reached on 571-272-1418. The fax phone number for the organization where this application or proceeding is assigned is 571-273-8300.

Information regarding the status of an application may be obtained from the Patent Application Information Retrieval (PAIR) system. Status information for published applications may be obtained from either Private PAIR or Public PAIR. Status information for unpublished applications is available through Private PAIR only. For more information about the PAIR system, see <http://pair-direct.uspto.gov>. Should you have questions on access to the Private PAIR system, contact the Electronic Business Center (EBC) at 866-217-9197 (toll-free). If you would like assistance from a USPTO Customer Service Representative or access to the automated information system, call 800-786-9199 (IN USA OR CANADA) or 571-272-1000.

/Robert Vetere/  
Examiner, Art Unit 1792

/Michael Cleveland/  
Supervisory Patent Examiner, Art Unit 1792